Extraction and Identification of Antioxidants from the Spice Aframomum danielli

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ABSTRACT: Successive extractions with diethyl ether and methanol of the whole seeds of the spice Aframomum danielli yielded diethyl ether extract (ADEE), 13.9%, and methanol extract (ADM), 3.4%, respectively. Similarly, reextraction of the defatted seeds of A. danielli successively with diethyl ether and methanol yielded extracts DFADEE (7.9%) and DFADM (6.7%), respectively. When these extracts were added to refined peanut oil (PNO) at 200 ppm, they showed good antioxidative effects. The percentage antioxidant effectiveness (AE) values were as follows: DFADM (87.3) > ADM (85.3) > ADEE (83.4) = tertbutyl hydroquinone (83.4) on day 20 of storage in an oven maintained at 65 ± 1°C. Generally, antioxidant extracts prepared from A. danielli were also more effective than butylated hydroxytoluene and α -tocopherol in stabilizing refined PNO. Antioxidant components of A. danielli were tentatively identified as phenolic compounds of the trihydroxy type with reducing properties. All extracts prepared from A. danielli showed strong ultraviolet-absorbing characteristics, and methanol was a good extracting solvent.

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KEY WORDS: Aframomum danielli, African spice, antioxidant effectiveness, lipid oxidation, natural antioxidants, phenolic compounds, protection factor.

Off-odor, off-flavor, and other deteriorative changes generally lead to product rejection and loss of revenue in the food industry. As far as fats and oils are concerned, the problems associated with lipid oxidation can be countered by the application of antioxidants. Synthetic antioxidants, such as butylated hydroxyanisole (BHA), butylated hydroxytoluene (BHT), and *tert*-butyl hydroquinone (TBHQ) are presently used in the food industry to stabilize oils against oxidation (1,2). However, adverse reports regarding these synthetic antioxidants exist in the literature (3–5). Therefore, in considering these reports and the present trends toward naturally preserved food products, herbs and spices have been targeted as sources of natural antioxidants (6–8). However, these reports involve several spices and herbs that are of commercial value

*To whom correspondence should be addressed. E-mail: aggk@nicfos.ernet.in (9), and there are few reports on the antioxidative activities of spices and herbs that have hitherto been of no commercial value. Therefore, the present work was carried out to isolate and identify the antioxidant compounds of the spice *Aframo-mum danielli*. *Aframomum danielli* belongs to the family Zingiberaceae, which is closely related to *Ammomum* Roxb. from tropical Asia. *Elletaria cardamomum* (L.) Maton, the source of commercial cardamoms, also belongs to the family Zingiberaceae (10).

MATERIALS AND METHODS

Spice and antioxidative screening procedure. The spice A. danielli was obtained locally in Ibadan, Nigeria. Prior to utilization, the seeds were cleaned of extraneous matter, pulverized, and sieved (60 mesh). Thereafter, the spice powder was added at 1, 2, 4, and 10% levels to refined peanut oil (PNO) (45 g) in 50-mL beakers. After careful mixing, they were transferred to an oven maintained at $65 \pm 1^{\circ}$ C for an accelerated storage study for up to 20-30 d. Control beakers contained only refined PNO, and some had synthetic antioxidants, BHT, or TBHQ (food-grade antioxidants). Peroxide values (PV) were determined every 5 d on the withdrawn samples by AOCS Official Method Cd 8-53 (11). The rate of oxidation was determined by using the procedure reported earlier (12). The percentage antioxidant effectiveness (AE) was calculated from the equation:

$$AE = \frac{PV \text{ of control} - PV \text{ of test sample}}{PV \text{ of control}} \times 100\%$$
[1]

The ability of *A. danielli* to stabilize PNO was also ascertained by using the oven weight gain method. The method described by Olcott and Einset (13) was modified as follows: 1.0 g sample of PNO (PV, 2.0 meq O_2/kg oil) was weighed into a clean 2-in. (50-mm) glass mycological Petri dish (Borosil Glass Works Ltd., Mumbai, India). Different amounts of petroleum ether extract of *A. danielli* powder (0.1, 1.0%), the defatted material (1, 10%), or synthetic antioxidant (Table 1) were then added to the Petri dishes, mixed carefully, and stored at 65 ± 1 °C as described before. Each sample was taken out of the oven for weighing every 24 h, cooled to room temperature (13), weighed, and returned to the oven. Increase in weight was recorded for up to 50 d, and

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TABLE 1 Monomolecular Rate Constant of Peanut Oil Treated with the Whole Spice Powder of Aframomum danielli^a

Treatment	Rate constant $(K_{\rm m} \times 10^3)$ at 65 ± 1°C
Refined peanut oil (PNO) (control)	$18.67 \pm 0.10^{\rm d}$
PNO + TBHQ, 0.02%	2.82 ± 0.20^{a}
PNO + A. danielli, 1.0%	$17.61 \pm 0.04^{\circ}$
PNO + A. danielli, 2.0%	$17.32 \pm 0.05^{b,c}$
PNO + A. danielli, 4.0%	$16.62 \pm 0.04^{\rm b}$
PNO + <i>A. danielli,</i> 10.0%	16.82 ± 0.11^{b}
$SE_m(12 \text{ df})$	± 0.21

^aValues are means \pm SD for three samples; any two means with different superscripts (a–d) are significantly different ($P \le 0.05$) SE_m = standard error of mean values; df = degrees of freedom.

the induction period was taken as the time required for the sample to gain 0.5% in weight to the nearest day. The protection factor (PF) was calculated from the equation:

$$PF = \frac{\text{day test attained } 0.5\% \text{ weight gain}}{\text{day control sample attained } 0.5\% \text{ weight gain}}$$
[2]

The data were subjected to two-way analysis of variance, followed by Duncan's new multiple range test for segregating differences between means (14).

Extraction of antioxidative components of A. danielli. The method described by Chang et al. (6) was modified as follows: Aframomum danielli powder (10 g) was refluxed with diethyl ether (100 mL) for 2 h at 60°C, cooled, and filtered. The filtrate was rotary-evaporated at 25°C (92–100 rpm/15 in. pressure, Rotavapor Model RE 121, Buechi Laboratoriums-Technik AG, Flawil, Switzerland) to get the antioxidant extract, ADEE. Similarly, the residue was successively reextracted with methanol (100 mL), allowed to cool to room temperature, filtered, followed by rotary evaporation at 40°C to yield another antioxidant extract, ADM. Residue obtained after fat extraction (i.e., after petroleum ether extraction for fat content determination, also known as the defatted material) (13.0 g) was also similarly reextracted with diethyl ether and methanol successively to yield antioxidant extracts DFADEE and DFADM, respectively. The extracts so obtained from A. danielli were added to refined PNO at 200 ppm, along with BHT and TBHQ as synthetic antioxidants, to get a comparative antioxidant effect assessment. Storage of treated PNO and periodic PV determinations were carried out as described earlier. Slower formation of peroxides was used as a measurement of the antioxidant activity of each extract (6).

Identification of antioxidant components. The methods described by Daniels and Martin (15) and Pratt and Miller (16) for identifying antioxidant activity [in accordance with the method of Duve and White (17)] was modified as follows: thin-layer chromatography (TLC) plates, laboratory-coated with silica gel G (Ranbaxy Laboratories Ltd., SAS Nagar, Punjab, India) to 0.3 mm thickness, were spotted with 20 μ L of each extract (from 1% solution in the respective solvent used for antioxidant extract preparation from *A. danielli*). The plates were then developed in the upper phase of chloroform/ethanol/acetic acid (98:2:2). Thereafter, the plates were sprayed with a solution of 9 mg of β -carotene dissolved in 30 mL chloroform, to which 20 mg of linoleic acid and 60 mL ethanol were added. The sprayed plates were exposed to sunlight of bright intensity (during the noon–afternoon hours) for 6 h. The retention of color of β -carotene (for antioxidant effect) or fading of orange color of β -carotene (for pro-oxidant or no antioxidant effect) were taken as corresponding to the relative antioxidative activity of each extract (18,19); TBHQ was used as a control antioxidant, which retained the orange color.

Furthermore, TLC tests to identify the chemical composition of the antioxidant extracts from A. danielli were carried out as follows: Spray 1: A 1% solution of potassium ferricyanide in water and a 1% solution of ferric chloride in water were sprayed on the developed TLC plates, which gave a blue color to indicate that these extracts may be phenolic compounds (20). Spray 2: Ferric chloride (2 g), dissolved in 100 mL ethanol, sprayed onto the developed TLC plates produced blue coloration for some spots, which indicates the probable presence of trihydroxy phenolic compounds (21). Spray 3: Ammoniacal silver nitrate solution was prepared by mixing 30 mL ammonium hydroxide and 70 mL water. Silver nitrate (3.4 g) in 100 mL water was added to the solution and sprayed on the developed chromatograms. After heating for 10 min at 105°C, brown, black, and gray spots were produced as evidence of reducing compounds (17). Separate TLC plates, developed in a similar fashion, were also exposed to iodine vapors to show the representative number of TLC spots in each extract.

Ultraviolet (UV)-visible absorption spectra of extracts were recorded on a UV-visible double-beam recording spectrophotometer (model UV-240; Shimadzu Corporation, Kyoto, Japan) from 0.0067% solutions of the extracts in methanol. Infrared (IR) spectra were also recorded on a Fourier-transform IR spectrophotometer (model Bruker IFS 25; Bruker, Karlshrue, Germany) with the Nujol mull technique in a sodium chloride cell over the range of $600-4000 \text{ cm}^{-1}$.

RESULTS AND DISCUSSION

Screening of A. danielli for antioxidative effect. Addition of whole ground powder of A. danielli at levels of 1, 2, 4, and 10% to refined PNO revealed only a marginal antioxidative effect of the spice compared to a synthetic antioxidant TBHQ (Table 1). Also, in the oven weight gain method, the petroleum ether extract of A. danielli and the defatted material did not show a good antioxidant effect, although it showed an induction period of 190 h and a PF of 0.79 (Table 2). Thus, prooxidant effects were found with the petroleum ether extract of A. danielli at 0.1 and 1.0% level and with 1.0 and 10% levels of the defatted powder of the spice when added to refined PNO (Tables 1 and 2). However, the pro-oxidant effect of A. danielli at high concentrations is not unusual because primary antioxidants are known to be effective at low concentrations, while at higher levels they may become pro-oxidants (22).

TABLE 2
Induction Periods of Peanut Oil Treated with the Petroleum Ether
Extract and Defatted Spice Powder of A. danielli

	Storage at 65 ± 1°C				
Treatment	Induction period (h)	Protection factor ^a	Remarks		
Refined PNO (control)	240 ± 17 ^p	1.0	_		
Refined PNO + BHT, 0.02%	313 ± 8^{q}	1.31 ± 0.07	Antioxidant		
Refined PNO + <i>A. danielli</i> extract ^b , 0.1% Refined PNO + <i>A. danielli</i>	167 ± 5 ^m	0.70 ± 0.04	Prooxidant		
extract ^b , 1.0%	$95 \pm 4^{ }$	0.40 ± 0.01	Prooxidant		
Refined PNO + <i>A. danielli</i> defatted powder ^c , 1.0%	238 ± 8 ^p	0.99 ± 0.07	No effect		
Refined PNO + A. danielli					
defatted powder ^c , 10.0%	190 ± 6^{n}	0.79 ± 0.04	Prooxidant		
SE _m (35 df)	± 4				

^aCalculated from the equation: Protection factor = (day test attained 0.5% weight gain)/(day control sample attained 0.5% weight gain). ^bPetroleum ether extract.

^cResidual powder after petroleum ether extraction of *A. danielli* powder. Values are means ± SD for six samples; any two means with different superscripts (l–q) are significantly different (P < 0.05). Abbreviations: SE_m, standard error of mean values; df, degrees of freedom; PNO, peanut oil; BHT, butylated hydroxytoluene; for other abbreviation see Table 1.

Antioxidant effects of A. danielli extracts. Extraction of the powder of A. danielli with different solvents, however, revealed the antioxidative effects of the spice. At 200 ppm, the order of percentage AE of the various extracts from A. danielli (Figs. 1–4) after 20 d, when added to refined PNO, at $65 \pm 1^{\circ}$ C was: DFADM (87.3) > ADM (85.3) > ADEE (83.4) = TBHQ (83.4). With BHT, the order was DFADM (87.3) > ADM (85.3) > ADEE (83.4) > BHT (79.6). The $K_{\rm m}$ values given in Table 3 indicate a significant antioxidant effect for ADM and DFADM, comparable to BHT and TBHQ. The effect of ADEE and DFADEE was considerably lower than for BHT and TBHQ, although a significant antioxidant effect was observed for the extracts compared to the control. To our

Monomolecular Rate Constant of PNO Treated with Diethyl Ether and Methanol Extracts of Whole Spice and Defatted Spice Powder of *A. danielli*

Treatment	Rate constant $(K_{\rm m} \times 10^3)$ at 65 ± 1°C
Refined peanut oil (PNO) (control)	$35.90 \pm 0.29^{\rm p}$
PNO + TBHQ, 0.02%	2.82 ± 0.20^{m}
PNO + BHT, 0.02%	$2.18 \pm 0.14^{l,m}$
$PNO + ADEE, 0.02\%^{a}$	7.83 ± 0.20^{n}
PNO + ADM, 0.02% ^b	1.08 ± 0.13^{1}
PNO + DFADEE, $0.02\%^{c}$	$10.59 \pm 0.23^{\circ}$
$PNO + DFADM, 0.02\%^d$	1.31 ± 0.20^{l}
SE _m (14 df)	± 0.44

^aADEE = diethyl ether extract of A. danielli.

^bADM = methanol extract of A. danielli.

^cDFADEE = diethyl ether extract of defatted powder of A.danielli.

^dDFADM = methanol extract of defatted powder of *A. danielli.* Values are means \pm SD for three samples; any two means with different superscripts (l–p) are significantly different (P < 0.05); SE_m = standard error of mean values; df = degrees of freedom. For other abbreviations see Table 2.

knowledge, this is the first report of antioxidant activity of *A. danielli*. The yields and colors of antioxidant extracts from the spice *A. danielli* by different solvents are shown in Table 4. A yield of 13.9% was obtained when *A. danielli* was extracted with diethyl ether (ADEE), but a lower yield (3.4%) was obtained with methanol (ADM). The extracts DFADEE and DFADM, prepared from the defatted material by successive extraction with diethyl ether and methanol, respectively, gave different yields and colors (Table 4).

The differences found for the antioxidant activities between whole spice petroleum ether extracts and of defatted material and those of diethyl ether and methanol extracts could be explained as follows. The antioxidant principles present in *A. danielli* are soluble in slightly polar solvents, such as diethyl ether and methanol. The antioxidant principles are not extracted by petroleum ether, and therefore, there is no



FIG. 1. Peroxide values of peanut oil treated with diethyl ether extract of *Aframomum danielli* (ADEE). BHT, sample treated with butylated hydroxytoluene; TBHQ, sample treated with *tert*-butyl hydroquinone.



FIG. 2. Peroxide values of peanut oil treated with methanol extract of Aframomum danielli (ADM). For other abbreviations see Figure 1.

antioxidant activity for this extract. However, for the defatted material, although the antioxidant is present, it is present inside the cellular compartments of the material, and the oil cannot reach inside and extract it. Also, for extraction of the antioxidant principles from the spice *A. danielli*, a nonpolar solvent such as petroleum ether (or a vegetable oil) may not be suitable. Diethyl ether and methanol, which are more polar solvents, extracted the antioxidant principles effectively, and therefore, there is a difference in antioxidant effect for the different extracts.

Tentative identification of antioxidant components. TLC patterns and R_f values of the components of antioxidant extracts from A. danielli are shown in Table 5. With R_f 0.0, the antioxidant components of ADM and DFADM were tenta-

tively identified to be phenolic compounds with reducing properties. The specific extinction coefficients in methanol for antioxidant extracts ADEE, DFADEE, ADM, and DFADM are shown in Table 6. The extracts had strong UV absorption at 228 nm (ADEE, DFADEE) and 210 nm (ADM, DFADM), and IR wavenumbers of 2560–4000 cm⁻¹ indicated possible phenolic hydroxyl groups in the compounds present in the extract, suggesting the presence of phenolic compounds in the extracts, similar to the specific spray reagents. According to Shahidi and Wanasundara (23), natural antioxidant extracts, such as ADEE, DFADEE, ADM, and DFADM from *A. danielli*, are primarily plant polyphenolic compounds that may occur in several parts of the plant. In line with our present work, additional studies have thus been



FIG. 3. Peroxide values of peanut oil treated with diethyl ether extract of defatted powder of *Aframomum danielli* (DFADEE). For abbreviations see Figure 2.



FIG. 4. Peroxide values of peanut oil treated with methanol extract of defatted powder of *Aframomum danielli* (DFADM). For abbreviations see Figure 2.

Yields and Colors of Antioxidant Extracts Isolated from A. danielli with Different Solvents						
Extract ^a	Solvent	Yield (% dry basis)	Appearance of extract	Starting material		
ADEE ADM DFADEE DFADM	Diethyl ether Methanol Diethyl ether Methanol	$13.9 \pm 0.10 \\ 3.4 \pm 0.20 \\ 7.9 \pm 0.10 \\ 6.7 \pm 0.15$	Reddish-yellow liquid Dark-brown powder Light-yellow powder Light-brown powder	A. danielli whole powder A. danielli whole powder A. danielli defatted material A. danielli defatted material		

^aFor abbreviations see Table 3.

TABLE 5

TABLE 4

Thin-Layer Chromatography Patterns, R_f Values and Tentative Identification of Antioxidant Components of A. danielli

	No.	R_{f}	Antioxidant	Phenolic	Trihydroxy	Reducing
Extract ^a	of compounds ^b	value	component ^c	component ^d	phenolic type ^e	component [*]
ADEE	11 (4)	0.43	+	+	_	_
ADM	2 (1)	0.00	+++	+	+	+
DFADM	4 (1)	0.00	+++	+	+	+
DFADEE	8 (2)	0.43	+	+	_	-

^aADEE, DFADEE, antioxidant diethyl ether extract from *A. danielli* whole powder and defatted material; ADM, DFADM, successive extracts obtained with methanol from *A. danielli* whole powder and from defatted material after diethyl ether extraction.

^bComponents found with iodine vapor, and number in parentheses is the compound number with the R_f value given alongside the table to show antioxidant activity.

^{*c*-*f*}Detected by specific spray reagents (Refs. 18–20).

TABLE 6

Ultraviolet Spectra	I Characteristics of	Antioxidant Extracts	from the Spice A. a	lanielli
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Extract ^a	A ^b (nm)	B ^c	C^d	А	В	С	D ^e (nm)
ADEE	206	1.48	113.9	228	1.64	126.2	228
DFADEE	206	1.00	149.9	220	1.11	166.4	228
ADM	210	2.34	350.8	282	0.69	103.5	210
DFADM	210	2.34	350.8	282	0.63	94.5	210

^aADEE, DFADEE: diethyl ether extracts from *A. danielli* whole powder and defatted material; ADM, DFADM: successive extracts obtained with methanol from *A. danielli* whole powder and defatted material after diethyl ether extraction.

^bA: wavelength of maximum absorption.

^cB: optical density.

^dC: extinction coefficient of 1% solution in 1-cm cell pathlength.

^eD: wavelength of maximum extinction coefficient.

carried out to identify natural phenolics that possess antioxidant activity (24–26).

Our present findings of the antioxidative effects of *A*. *danielli* should therefore rekindle more research into the application of natural antioxidants for lipid stabilization.

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REFERENCES

- 1. World Health Organization, *Food Additive Series No. 3*, World Health Organization, Geneva, Switzerland, 1972.
- 2. Giese, J., Fats, Oils and Fat Replacers, *Food Technol.* 50:78–84 (1996).
- 3. Branen, A.L., Toxicology and Biochemistry of Butylated Hydroxy Anisole and Butylated Hydroxy Toluene, *J. Am. Oil Chem. Soc.* 52:59–63 (1975).
- Ito, N., M. Hirose, S. Fukushima, H. Tsuda, T. Shirai, and M. Tatematsu, Studies on Antioxidants: Their Carcinogenic and Modifying Effects on Chemical Carcinogenesis, *Food Chem. Toxicol.* 24:1071–1082 (1986).
- Takahashi, O., Haemorrhages Due to Defective Blood Coagulation Do Not Occur in Mice and Guinea Pigs Fed Butylated Hydroxy Toluene but Nephro Toxicity Is Found in Mice, *Ibid.* 30:89–97 (1992).
- Chang, S., B. Ostric-Matijasevic, O.A.L. Hsieh, and C. Li Huang, Natural Antioxidants from Rosemary and Sage, *J. Food Sci.* 42:1102–1106 (1977).
- Jitoe, A., T. Masuda, I.G.P. Tengah, D.N. Suprapta, I.W. Gara, and N. Nakatani, Antioxidant Activity of Tropical Ginger Extracts and Analysis of the Contained Curcuminoids, *J. Agric. Food Chem.* 40:1337–1340 (1992).
- Madsen, H.L., and G. Bertelsen, Spices as Antioxidants, Tr. Food Sci. Technol. 6:271–277 (1995).
- 9. Farrell, K.T. (ed.), *Spice, Condiments and Seasonings,* AVI Publishing Company, Westport, 1990, p. 224.
- Lock, J.M., Zingiberaceae, in *Flora of Tropical East Africa*, edited by R.M. Polhill and A.A. Balkema, Rotterdam, 1985, p. 1.
- 11. Official and Tentative Methods of the American Oil Chemists'

Society, 4th edn., American Oil Chemists' Society, Champaign, 1990, Official Method Cd 8-53.

- Gopala Krishna, A.G., and J.V. Prabhakar, Influence of Water Activity on Secondary Products Formation in Autoxidizing Methyl Linoleate, J. Am. Oil Chem. Soc. 69:178–183 (1992).
- Olcott, H.S., and E. Einset, A Weighing Method for Measuring the Induction Period of Marine and Other Oils, *Ibid.* 35:161–162 (1958).
- Amerine, M.A., R.M. Pangborn, and E.B. Roessler (eds.), *Principles of Sensory Evaluation of Food*, Academic Press, New York and London, 1965, p. 437.
- Daniels, D.G.H., and H.F. Martin, Antioxidants in Oats: Monoesters of Caffeic and Ferulic Acids, J. Sci. Food Agric. 18:589–595 (1967).
- Pratt, D.E., and E.E. Miller, A Flavonoid Antioxidant in Spanish Peanuts (Arachis hypogaea) J. Am. Oil Chem. Soc. 61:1064–1067 (1984).
- Duve, K.J., and P.J. White, Extraction and Identification of Antioxidants in Oats, *Ibid.* 68:365–370 (1991).
- Marco, G.J., A Rapid Method for Evaluation of Antioxidants, *Ibid.* 45:594–598 (1968).
- 19. Taga, M.S., E.E. Miller, and D.E. Pratt, Chia Seeds as a Source of Natural Lipid Antioxidants, *Ibid.* 61:928–931 (1984).
- Barton, G.M., R.S. Evans, and J.A.F. Gardner, Paper Chromatography of Phenolic Substances, *Nature* 170:249–250 (1952).
- Reio, L., A Method for the Paper-Chromatographic Separation and Identification of Phenol Derivatives, Mould Metabolism and Related Compounds of Biochemical Interest Using Reference System, J. Chromatogr. 1:338–373 (1958).
- 22. Rajalakshmi, D., and S. Narasimhan, Food Antioxidants: Sources and Methods of Evaluation, in *Food Antioxidants, Technological, Toxicological and Health Perspectives*, edited by D.L. Madhavi, S.S. Deshpande, and D.K. Salunkhe, Marcel Dekker Inc., New York, 1996, pp. 65–157.
- 23. Shahidi, F., and P.K.J.D.P. Wanasundara, Phenolic Antioxidants, CRC Crit. Rev. Food Sci. Nutri. 32:67–103 (1992).
- Su, J.-D., T. Osawa, S. Kawakishi, and M. Namiki, Tannin Antioxidants from *Osbeckia chenensis*, *Phytochemistry* 27: 1315–1319 (1988).
- 25. Igile, G.O., W. Oleszek, M. Jurzysta, S. Burda, M. Fafunso, and A.A. Fasanmade, Flavonoids from *Vernonia amygdalina* and Their Antioxidant Activities, *J. Agric. Food Chem.* 42:2445–2448 (1994).
- Tsuda, T., K. Oshima, S. Kawakishi, and T. Osawa, Antioxidative Pigments Isolated from the Seeds of *Phaseolus vulgaris* L., *Ibid.* 42:248–251 (1994).

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